

# Application of Zeolite, A Biomaterial Agent, in Dental Science: A Review Article

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**Introduction:** Zeolite is an aluminosilicate biomaterial which has been used widely in medicine, tissue engineering, drug delivery systems, wound dressing, as an antibacterial agent and also it has been taken into consideration in dental sciences. So the aim of this study is reviewing the application of this amazing substance in dental sciences. **Materials and Methods:** In recent years there are many investigations on application of Zeolite. This review has been done on the application of this agent in dentistry. According to investigations from 1980, zeolites applications in dentistry were classified to the application in tissue engineering, root canal therapy, prosthodontics, periodontics and implant, restorative dentistry and oral medicine. **Results:** zeolite is a porous, biocompatible and antibacterial agent and according to its suitable properties it has been used as tissue engineering scaffold, wound dressing and so on. Zeolite is an antibacterial agent that can be used in root canal disinfection and in soft liner dentures. It can inhibit pathogens of dental caries and periodontal disease. According to biocompatibility of Zeolite, it also can be used as implant coating to improve osteointegration. It has also been used to detect squamous cell carcinoma. **Conclusion:** It seems that Zeolite can be considered as an amazing substance in dental sciences.

**Keywords:** Zeolite; Tissue engineering; Wound dressing; Dentistry

## Introduction

Zeolite is a porous aluminosilicate agent with pore size between 3 to 10 Å in diameter. These pores can trap cations (Ag, Zn), water, and organic molecules and exchange them by other cations. By this way zeolite can display appealing features such as antimicrobial effect (1, 2). Its main constituents are phosphorus, aluminum, oxygen, silicon, etc., which they are repeated in the structure (3). Zeolites are divided into synthetic and natural groups. Zeolites are widely used in various sciences. zeolite has been used in industry (4), agriculture (5), Animal feed and poultry (6, 7), Water pollution treatment (8, 9), biochemistry (10) and so on. Recently it was taken into consideration in medicine. They are widely used in medicine such as wound dressing (11, 12), drug delivery (13, 14), as an antibacterial agent (15, 16), anticancer agent (17-20), for osteoporosis treatment (21, 22) treatment of various diseases (23, 24), tissue engineering (25) and so on. Zeolite has been also used in dentistry especially for antibacterial properties. The aim of this study is reviewing the application of the zeolite in dentistry.

## Materials and Methods

Search has been done by the keywords of zeolite, tissue engineering, wound dressing and dentistry in PubMed and Google scholar search engines for finding all of the available evidence from 1980.

## Results

In addition to the diverse and extensive applications of zeolite in various sciences, there is a great attention in dental sciences as well. The applications of this agent were classified in regenerative dentistry, tissue engineering and wound healing, root canal therapy, prosthodontics, oral medicine, restorative dentistry, periodontics and implant and so on as further discussed below.

### Regenerative dentistry

Today tissue engineering is very much considered in various medical fields, as well as in the soft and hard tissue engineering of the oral cavity. Zeolite is one of the materials used as a scaffold in bone tissue engineering due to its

chemical and porous structure and the possibility of implantation of cells in these pores. Pazarçeviren *et al.* investigate the clinoptilolite/poly ( $\epsilon$ -caprolactone)-poly (ethylene glycol)-poly ( $\epsilon$ -caprolactone) composite scaffolds as bone tissue engineering scaffold. Clinoptilolite improve the mechanical properties of the scaffold. The scaffold showed higher osteoinductivity in vitro (26). According to Costa *et al.*  $\alpha$ -PVDF / NaY-32, which contains Zeolite NaY as filler, increases osteoblast and fibroblast proliferation. In addition, this scaffold did not show a significant pro-inflammatory effect. Therefore, zeolite has favourable properties for use as a tissue engineering scaffold (27). Schutze *et al.* indicated that zeolite-A reduces the number of pits per osteoclast and also reduces the activity of the cathepsin B enzyme by 3 times. So it seems that zeolite can inhibit bone resorption (22). Keeting *et al.* indicated that Zeolite increases mRNA of growth factor beta 1 levels and induces proliferation and differentiation of osteoblast (28). According to Chen *et al.* nano zeolitic imidazolate framework-8 Loaded onto titanium surface, in addition to the anti-bacterial property against Streptococcus mutans, increased the activity of alkaline phosphatase, extracellular matrix mineralization and osteogenic genes expression in MG63 cells (29).

Zeolite is also used as wound dressing to improve healing process. Salehi *et al.* evaluated the nanozeolite / starch thermoplastic hydrogels that were loaded by chamomile extract. The release of drug in the wound containing zeolite was more sustained. The MTT test indicated the biocompatibility of this wound for fibroblast mouse cells (L929). Histological assessment on the rats revealed that the wound dressing leads to improved epithelialization, collagen formation, and angiogenesis and also reduces inflammation. In addition the pilot study of five patients with chronic ulcers by car accident, bed sore, pressure ulcer, burn trauma and burn ulcer indicate improvement in wound healing without any hypersensitivity or other complications (11). Neidrauer *et al.* evaluate a topical ointment containing nitric-oxide-loaded zeolites. This agent showed antibacterial properties against Gram-negative bacteria (*Escherichia coli* and *Acinetobacter baumannii*), Gram-positive bacteria (*Staphylococcus epidermidis* and methicillin-resistant *Staphylococcus aureus*) and fungus (*Candida albicans*). In addition, it was biocompatible for 3T3 fibroblast cells. It also accelerated wound healing in rats (30). Ninan *et al.* investigated pectin / copper exchange faujasite hybrid membrane. The membrane indicated thermal stability, antibacterial activity, biocompatibility for NIH3T3 fibroblast and improvement in

wound healing and re-epithelialisation in rats (31). According to Ninan *et al.* gelatin/copper activated faujasites was biocompatible for NIH 3T3 fibroblast cells. It also improves wound healing in rats (32). Ninan *et al.* also investigate gelatin/hyaluronic acid (HA)/faujasite porous scaffolds. It was biocompatible for NIH 3T3 fibroblast cell lines and promotes wound healing in rats (33). Seifu *et al.* indicate that Loading fluorinated-zeolite (FZ) particles on polyurethane scaffolds increase oxygen delivery to Human coronary artery smooth muscle cell, improve cell proliferation and infiltration depth in scaffold (34). According to zeolite appropriate properties it seems that it can be considered as suitable mucosal wound dressing as well.

### Root canal therapy

Recently, zeolite solution has been considered as an agent for canal irrigation due to its antibacterial properties. In the study of Ghivari *et al.* the antibacterial properties of silver-zeolite on bacteria *E. faecalis*, *Staphylococcus aureus*, *C. albicans*, which lead to failure of root cultures, at intervals of 1 s, 5 s, 10 s, 30 s, and 60 s, were evaluated by simulating the biofilm model. In this study Silver-zeolite 2% indicated the least antibacterial property against *E. faecalis*, *S. aureus*, and *C. albicans* compared to NaOCl (5.25%), 2% CHX, OCT (0.10%) (35). Probably the anti-bacterial property of silver-zeolite is related to silver to release or oxygen Increase (36). Since zeolite can trap cations such as silver in its pores of crystalline structure, it can release them gradually. Perhaps because of the fact that zeolite-silver has been investigated as irrigant in this study and therefore its crystalline structure is clogged, it has shown little antibacterial property. Odabas *et al.* evaluated the effect of adding 0.2% and 2% silver-zeolite to the mineral trioxide aggregate (MTA) in antibacterial properties against *Staphylococcus aureus*, *Enterococcus faecalis*, *Escherichia coli*, *Pseudomonas aeruginosa*, *Candida albicans*, *Porphyromonas gingivalis*, *Actinomyces israelii*, and *Prevotella intermedia* on agar plates at 24, 48, and 72 h time points. MTA incorporated silver-zeolite 0.2% and 2%, showed antibacterial properties on all microorganisms except *P. intermedia* and *A. israelii*, at all time intervals. While MTA without silver-zeolite showed antibacterial effect only against *C. albicans*, *E. coli*, and *P. intermedia*. Therefore, the addition of zeolite-silver increased the antibacterial properties of MTA (37). Torabinejad *et al.* reported that MTA has no effect on *E. faecalis* and *S. aureus* (38). Estrela *et al.* also observed that MTA has no effect on *E. faecalis* (39). Considering that some studies have shown the antibacterial properties of MTA in relation to high PH (40-43), It has been shown that PH score

increases by adding zeolite-silver up to 11 and the antibacterial properties of MTA will enhance subsequently (37). Ghatole *et al.* In 2016 evaluate the anti-bacterial property of 2% silver zeolite added to calcium hydroxide mixed with sterile water and calcium hydroxide mixed with 2% chlorhexidine against *Enterococcus Faecalis* in brain-heart infusion (BHI) broth by spectrophotometer on day 1, 3 and 7. They reported that in all three time intervals, 2% silver zeolite added to calcium hydroxide mixed with sterile water had the highest inhibitory effect (44). In another study, Ghatole *et al.* evaluated the antibacterial properties of MTA, MTA + Silver-Zeolite, MTA + Chlorhexidine by spectrophotometrically using an ELISA-reader in the suspension of *Enterococcus faecalis* on the test materials. The results indicated that all three materials had antibacterial properties at 1, 3 and 7 time intervals. MTA with silver zeolite showed the highest antibacterial properties, While MTA showed the least antibacterial properties (45). Other studies also found that MTA was ineffective on *E. faecalis* (46, 47). Others also report the effect of MTA on *E. faecalis* (41). Cinar *et al.* added Silver zeolite (SZ) at 0.2 and 2% to glass ionomer cement. 2% SZ containing GIC had the highest inhibitory effect on all groups including *Streptococcus milleri*, *Staphylococcus aureus*, and *Enterococcus faecalis* after 24 and 48 h (48). Patel *et al.* showed that zeolites can increase the antibacterial properties of glass ionomer cement, which was observed in vitro for up to 90 days (49). Thom *et al.* compare various sealers including ZUT, Ketac-Cem (GIC component of ZUT), Ketac-Endo and AH 26 sealer for cytotoxicity. In terms of hemolytic and cytotoxic, ZUT was similar to GIC and better than AH26 (50). ZUT is a GIC combined with an antimicrobial zeolite that is made at the University of Toronto's Faculty of Dentistry (49). In the other hand physical properties of zeolite has been considered. In the evaluation of compressive strength by ISO 9917-1, this value was lower for MTA + 2% Ze-Ag-Zn than MTA at day 4 and 21, but MTA + 2% Ze-Ag-Zn did not show cytotoxicity for pulmonary adenocarcinoma cells by MTT test. Therefore, the use of MTA + Ze-Ag-Zn is appropriate in cases where compressive strength is not high (51). In another study comparing MTA and MTA with 0.2% or 2% silver zeolite, it was shown that 2% SZ MTA released the highest calcium value at 24 h and the setting time was decreased compared to MTA. Also, 2% SZ MTA had a higher solubility than MTA and 0.2% SZ MTA at day 7 (52). Lead zeolite and zinc zeolite shorten the setting time of Portland cement. (53) In the Lalh *et al.* study, ZUT also showed good dentin bonding (54). Zinc Zeolite improved the dentin shear bond strength (55).

### Prosthodontics

Nikawa incorporated silver zeolite for the first time in a tissue conditioner which led to proper plaque control and also an inhibitory effect on *Candida albicans* (1). Tissue conditioners containing silver zeolite were able to inhibit *Candida albicans* as well as *Staphylococcus aureus* and *Pseudomonas aeruginosa* bacteria in saliva immersed media for four weeks. Also it has been shown that tissue conditioners containing silver zeolite can inhibit *Candida albicans* and *Staphylococcus aureus*, methicillin-resistant *S. aureus* (MRSA) over a period of 28 days, and this property did not change with immersion in saliva (56). Saravanan *et al.* evaluated the viscoelastic properties of soft liner containing zeolite after 28 days application by patients. Silver- zeolite could inhibit *candida albicans*, besides did not change the viscoelasticity in comparison to control group (57). Although PMMA is strictly used in the manufacture of dental prostheses, the probability of bacterial growth is high (58-60). Silver zeolite can have antibacterial properties as Biocide-releasing polymers, while no tissue irritations have been reported (37). Silver zeolite has been inhibited in *C. albicans* by in vitro, but in the presence of saliva coat, its inhibitory effect was reduced (1). Casemiro *et al.* evaluated the effect of adding silver zinc zeolite in denture base resins at 2.5%, 5.0%, 7.5%, and 10%. It has been concluded that there is antimicrobial activity against *C. albicans* and *S. mutans*, but adding high concentrations of zeolite has a negative effect on mechanical properties (61). Silver zeolite can be further evaluated as a substance for optimization of the PMMA-antimicrobial formulation (62). Self-cured acrylic resins are used in the prosthesis temporary crowns or repairing fractured prosthesis. Self-cured acrylic resins can become contaminated and create a very bad smell, if it remain for long time. Also, bacteria found in periodontal diseases have been found in culture of the temporary crowns. It has been also seen that by addition of a resin sample to the *Candida albicans* culture, it could grow over a short period of time (63). Casemiro *et al.* assessed the antibacterial effect of acrylic resins containing different percentages of silver and zinc zeolite. It indicated that the addition of silver-zinc zeolite could inhibit *Candida albicans* and *Streptococcus mutans*, but concentrations of 2.5% and higher silver-zinc zeolite resulted in reducing flexural strength. (61) Kuroki *et al.* evaluated antibacterial properties as well as the amount of residual monomer by adding antibacterial agents to UNIFAST III as self-cured acrylic

resins. For this purpose, Zeomic (Carrier: A type zeolite), NOVARON (Carrier: Zirconium phosphate), Bactekiller (Carrier: Zirconium phosphate) were added to UNIFAST III as an inorganic antibacterial agent and their impact on *Streptococcus mutans* were examined. *S. mutans* attachment was significantly lower in all three groups than in the control group. Residual viable count of *S. mutans* (CFU) was significantly decreased in all three groups compared to controls. Residual monomer content was similar to control group. Therefore, by adding an inorganic antibacterial agent, antibacterial properties can be improved in self-cured acrylic resins (64). Adding silver zeolite does not affect the inherent dynamic viscoelastic properties of tissue conditioner's (65).

### Oral medicine

Oral cancer is the sixth most common cancer in the world, most of which is oral squamous cell carcinoma. Recently, volatile organic compounds (VOCs) discharged from human body have been considered as a way of detecting pathologies. Many studies have shown that VOCs, as molecular cancer markers, are produced by in vitro cancer cell lines. (66-70) Shigeyama *et al.* in 2018, evaluated VOCs in patients with oral squamous cell carcinoma by a method of combining thin-film microextraction based on a ZSM-5 / polydimethylsiloxane hybrid film coupled with gas chromatography-mass spectrometry. 35 VOCs were similar between the patient and the control group. In addition, 7 VOCs in the healthy group and 38 VOCs were seen in the affected group and the release pattern of these VOCs was evaluated in two groups. This study showed that ZSM-5 / polydimethylsiloxane hybrid film can be used to identify tumor-specific candidate biomarkers (71). According to the release of VOCs into the saliva through simple diffusion of blood (72), and the fact that saliva collection is easier, less invasive and more cost effective than getting blood sample (73-75), it seems that zeolite can be considered as a tumour marker in early detection of cancer.

### Restorative dentistry

Silver zeolite can inhibit bacteria responsible for dental caries including *Streptococcus mutans*, *Streptococcus sanguis* and *Actinomyces viscosus* (76). Can-Karabulut *et al.* investigated the in vitro dentin bond strength of a composite resin after the removal of zeolite as temporary cement after 7 days. Shear bond strength for the composite material was significantly lower in the group that used zeolite as temporary cement in comparison to the control group which no cement was used (77).

### Periodontics and implant

Kawahara *et al.* Showed that silver-zeolite can inhibit *Porphyromonas gingivalis*, *Prevotella intermedia* and *Actinobacillus actinomycetemcomitans* as major periodontopathogens in anaerobic condition. Therefore It can be used in anaerobic conditions, including periodontal packets (76). It has also been shown in other studies that Silver zeolite has antibacterial effect on oral pathogens responsible for periodontal diseases and caries, including *S. mutans*, *S. mitis*, *C. albicans*, *S. aureus* and *P. aeruginosa* in aerobic conditions in vitro (2, 78). Also, silver zeolite suspended in PBS as mouth rinse was shown to reduce supra-gingival plaque formation (79). Silver-zeolite has antibacterial effect on *Streptococcus mutans*, *Lactobacillus casei*, *Candida albicans* and *Staphylococcus aureus* in vitro. So it can be considered as way of Oral hygiene control (80). silver-zeolite inhibit *P. gingivalis* and this effect was stronger in comparison with silver alone. silver-zeolite and a polyphenol-rich extract of *A. nodosum* (ASCOP) has antibacterial effect on *P. gingivalis*, reduce TNF $\alpha$  and IL-6 secretion and it has antioxidant properties. Therefore it can be effective against periodontal disease (81). The silver-and zinc-containing zeolite matrix (AgION) was used as a coating for stainless steel and was shown to be antibacterial against *Escherichia coli*, *Staphylococcus aureus*, *Pseudomonas aeruginosa*, and *Listeria monocytogenes* (82). In the study by Li *et al.* ZSM-5 zeolite was used as a coating on titanium components. In an animal study on rabbits, more bones were formed on zeolite-coated implants (83). Coating the MFI zeolite on a titanium alloy surface improves biocompatibility, corrosion resistance, osteoconductivity and osteoinductivity in vitro (84).

### Discussion

Zeolites are biocompatible and high-quality materials for carrying various agents, including drugs and antibacterial agents and release them gradually. In addition to vast applications in medical and other sciences, it also has high potential of use in the field of dentistry. Zeolite is a potential wound dressing and tissue engineering scaffold in oral cavity. Zeolite is used in dentistry with silver for antimicrobial applications inside the root canal and in soft liner dentures and so on. Zeolite can inhibit pathogens of dental caries and periodontal disease. In addition zeolite indicated good biocompatibility properties for osteoblasts as implant surface coating. It has also been used to detect squamous cell carcinoma.



## Conclusion

According to the attractive properties of zeolite, it can be concluded that Zeolite has the potential of variety application in dental sciences, especially in combination with antibacterial agents.

Conflict of Interest: 'None declared'.

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